



Beneficial Uses of CO₂

Technologies and Implications for a Carbon Market

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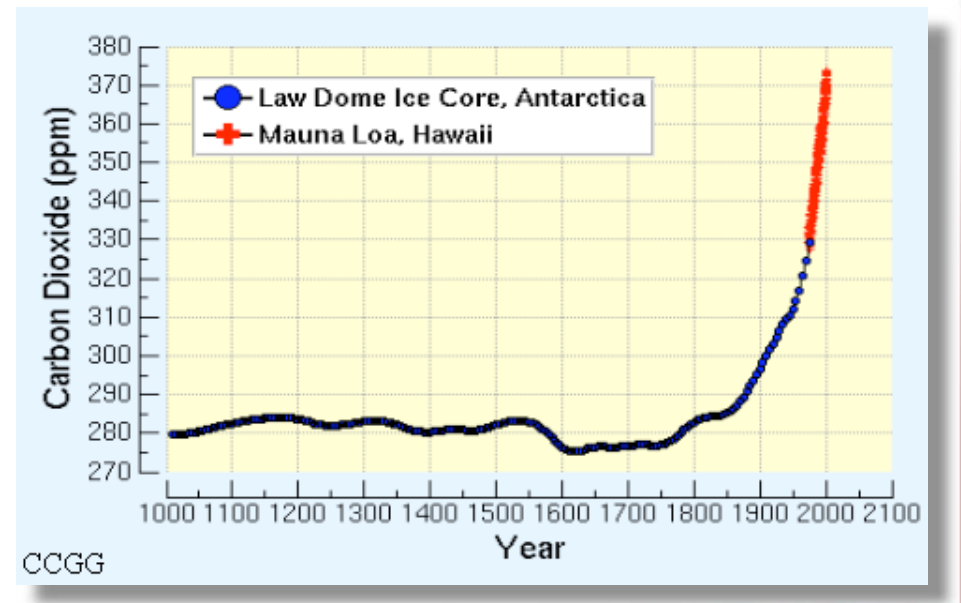
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Introduction

- Average preindustrial level = 280 ppm (~2200 Gt)
- Human activity currently adds 2 ppm/yr (15 Gt /yr)
- Current total atmospheric CO₂ = 380 ppm (~ 2900 Gt)
- Primary sources:
 - Electricity production
 - Vehicular transport



Primary sources of CO₂

Source	US 2000 (Gt)
Fossil fuel combustion	5.6
[Electricity generation]	[2.2]
[Transportation]	[1.9]
Iron and steel production	0.066
Cement manufacture	0.041
Waste combustion	0.022
Ammonia manufacture	0.018
Lime manufacture	0.013
Limestone/dolomite use	0.009
Natural gas flaring	0.006
Aluminum production	0.005
Soda ash manufacture/consumption	0.004

Source: EPA, EIA and SPCC TAR



CO₂: Pollutant, By-product or Waste?

- Carbon trading, carbon taxes, or other forms of carbon control will put a value on CO₂ as a GHG
- Other waste streams have been found to have value
 - Aluminum cans, glass containers, newspapers, cardboard, etc.
- Four strategies linked to the value of CO₂
 - Reduce
 - **Displace other GHG**
 - Sequester
 - **Convert and reuse**



Global Warming is More Than Just CO₂

Gas	GWP
Carbon Dioxide	1
Methane	23
Nitrous Oxide	296
HFC-23	12,000
HFC-125	3,400
HFC-134a (automotive A/C)	1,300
HFC-143a	4,300
HFC-152a	120
HFC-227ea	3,500
HFC-236fa	9,400
Perfluoromethane (CF ₄)	5,700
Perfluoroethane (C ₂ F ₆)	11,900
Sulfur Hexafluoride (SF ₆)	22,200

Refrigerants {

Source: ICPP Third (2001) Assessment Report

GWP = Global warming potential. GWP(CO₂) = 1



Displacement: CO₂ Replaces Higher-Impact Greenhouse Gases

- An Example: CO₂ as a replacement for HFC-134a refrigerant commonly used in vehicles
 - HFC-134a GWP = 1300
- Average leakage = 40 g/year/vehicle
- 243 million vehicles in US (2004), ~ 90% have A/C → ~ 0.013 Gt/yr CO₂ equivalent emitted
- CO₂ can be a solution, not a problem



Reuse: Delay the Release of CO₂ to the Atmosphere

Present Primary Industrial Uses of CO₂

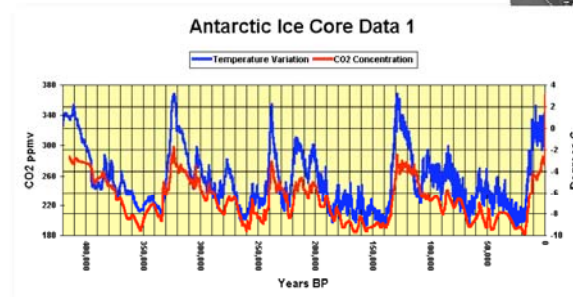
	Market (Gt/yr)	Lifetime
Fertilizer (urea)	0.090	Six months
Methanol	0.024	Six months
Inorganic carbonates	0.008	Decades to centuries
Organic carbonates	0.003	Decades to centuries
Polyurethanes	0.010	Decades to centuries
Technological	0.010	Days to years
Food	0.008	Months to years

“A large proportion of all CO₂ recovered is used at the point of production to make further chemicals of commercial importance, chiefly urea and methanol” *IPCC report 2005*



Convert and Reuse: *Only Fuel Scales to CO₂ Problem*

- ‘Problem’ CO₂ came from fuels
- In principle, it can be recycled
- BUT, where does the energy come from to convert CO₂ back to fuel?
 - Solar (biofuels... or something better)
 - Nuclear



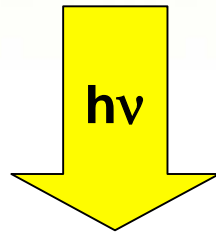
How to capture the CO₂?

- Two major possibilities
 - Capture it at the source
 - Most practical for stationary sources
 - Easiest with pure oxygen combustion
 - Demonstrations now underway
 - Remove it from the atmosphere
 - Challenging, but not impossible
 - Wind naturally moves vast quantities of air
 - Feasible to build scrubbers that to pull CO₂ directly from air
 - Potential to disconnect capture from source
 - Not yet demonstrated at scale or in field

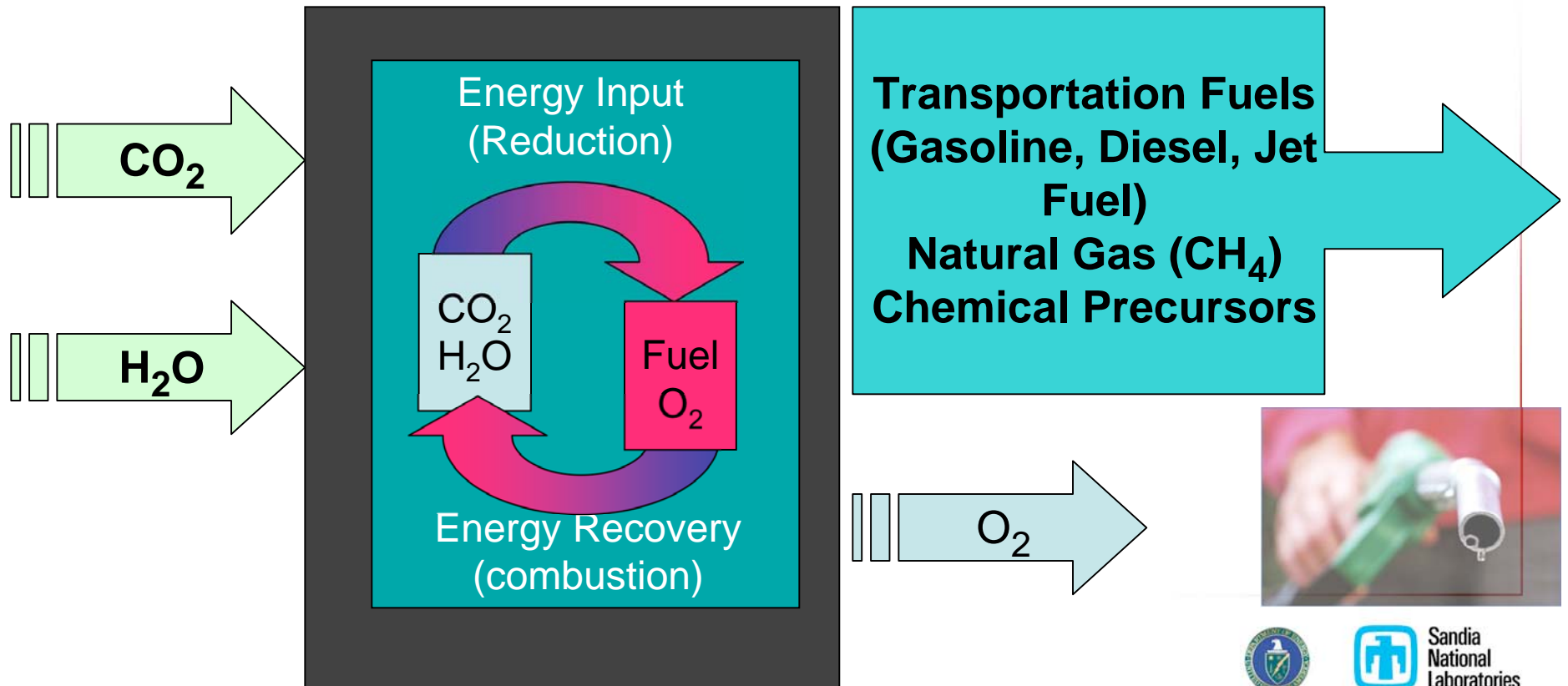


Solar Driven CO₂ to Fuel

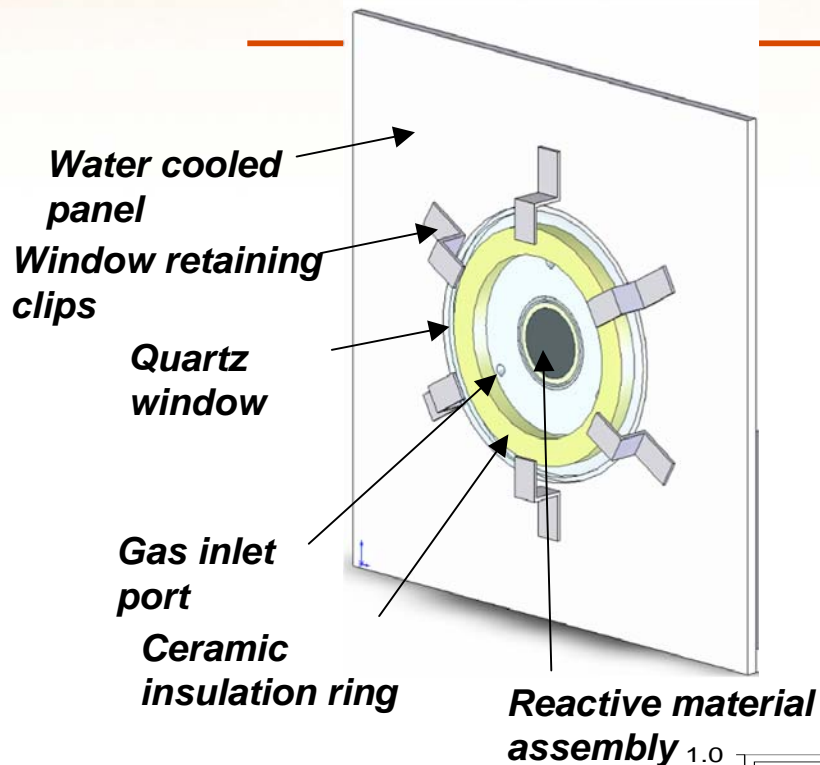
Biofuels Without the Biology



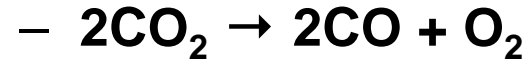
Solar Heating



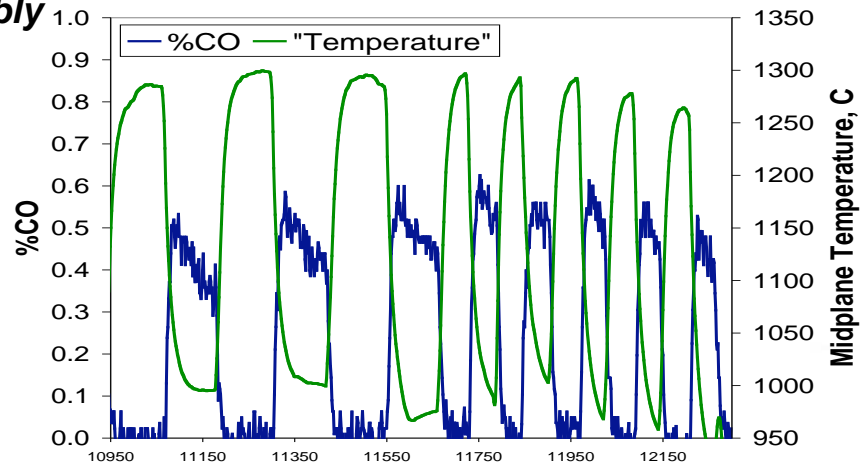
CO₂ Splitting Shown Feasible



- Two types of catalysts have demonstrated the ability to split both H₂O and CO₂:




- H₂ and CO combine to produce Syngas
- Syngas can be converted to any hydrocarbon fuel



Technically Possible: Moving CO₂ From Waste to Resource

- Technology development needed
 - CO₂ capture development
 - CO₂ to fuel efficiency improvements
- Regulation needed
 - True cost of CO₂ occurs on a timescale incompatible with market forces
 - Regulation is needed to connect these costs on a more immediate timescale





Market Questions

- How does capture, convert and reuse change the carbon market?
- How long a delay must be realized to earn a carbon credit?
- Can today's market mechanisms spur investment in the recycle of CO₂?

